

## ILARA MOKIN, ONDO STATE

## FACULTY OF BASIC AND APPLIED SCIENCES

## DEPARTMENT OF PHYSICAL AND CHEMICAL SCIENCES FIRST SEMESTER 2017/2018 B,Sc EXAMINATIONS

## BCH 201- GENERAL BIOCHEMISTRY I

Time:	2 hrs	
		CAA BONG

- Instruction: Attempt any (4) of the six (6) questions
- 1. (a) Define pH
  - (b) Why is pH important in Biology?
  - (c) How can you determine the pH of a given solution?
  - (d) Calculate the pH:
    - (i) of 0.5M sodium hydroxide solution?
    - (ii)) If 50ml of 0.32M NaOH are added to 100ml 0.2M HCl

(15 marks)

- 2. (a) What are buffers?
  - (b) Define Buffer Capacity
  - (c) At what point does a buffer works best?
  - (d) From the reaction,  $HA \leftrightarrow H^+ + A^-$ ; Derive  $pH = pKa + Log [A^-]/[HA]$ .
  - (e) A buffer solution contains equal concentrations of acetic acid and sodium acetate. Calculate the pH of the buffer.

    (15 marks)
- 3. (a) Giving correct examples, write short notes on
  - (i) Epimers...... 3 marks
  - (ii) Enantiomers..... 3 marks
  - (iii) Anomers......3 marks
  - (b) Clearly state the similarities and differences between Amylose and Amylopectin....6 marks
- 4. (a) With specific examples, differentiate between
  - (i) Purines and Pyrimidines...... 3 marks

  - (iii) Monocistronic and Polycistronic gene..... 3 marks
  - (b) Describe

    - (iii) Cruciform......2 marks
- 5. (a) State clearly the similarities and differences between enzymes and catalyst...5 marks
  - (b) With the aid of annotated diagrams only, describe

	(b) Wi	th the ai	d of annotated diagrams only, describe
		(i)	Enzyme 2 marks
		(ii)	Substrate2 marks
		(iii)	Ezyme-substrate complex 2 marks
	(c)	Briefly	define any four (4) of the following:
		(i)	Active site1 mark
		(ii)	Binding site1 mark
		(iii)	Activation energy 1 mark
		(iv)	Cofactors 1 mark
		(v)	Coenzymes 1 mark
(l	(a) Cla	assify the	e 20 amino acids based on the side chains attached to the α-carbon7 marks
	(b) Illu proteir	ıstrate w	vith suitable diagrams and examples, the four levels of structural organization of